

Structural Characterization of Fe₃O₄ Nanoparticle Embedded Polypyrrole-FeCl₃ Composite

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Abstract: *The conducting polymer composites have generated lot of scientific interest and led to active multi-disciplinary research because of their excellent potential for technological applications. Among the conducting polymers, Polypyrrole (PPY) and its composites have attracted considerable attention because they are easily synthesized. There are many approaches in the enhancement of the mechanical strength, chemical stability and gas-sensing properties by combining PPY with organic and inorganic materials to form composites. The composite of the various combinations of ratios the Pyrrole to FeCl₃ with the Fe₃O₄ nano particles. The composite prepared in the form of powder. The powder of composite is prepared through Chemical polymerization method. The sample preparation starts from distillation of Pyrrole with various combination of Pyrrole to FeCl₃ are stirred for overnight. The distillation of Pyrrole for purification of Pyrrole, we have designed the Portable Distillation Set up which 100% save the wastage of water used for the cooling of vapor in Condenser. Also the Fe₃O₄ nano particles are added for the blending purpose. The solution is then processed in centrifugal machine and filtered to separate the composite particles from the solution. The powder is dried in the furnace at warm temperature. Also the composite is exposed to room temperature for sufficient time. The dried powder sample is analyzed through XRD.*

Keywords: PPY, XRD

I. INTRODUCTION

The key property of most polymers, which distinguishes them from metals, is their inability to carry electricity. Whereas, the insulating properties of polymers are a significant advantage for many applications of plastics. During the past 25 years, however, a new class of organic polymers has been devised with the remarkable ability to conduct electrical current. Part of a larger class of materials called “synthetic metals”. Some of these conductive plastics are already under development for practical applications. A major obstacle to the rapid development of conductive polymers has been the lack of understanding of how electrical conductivity works in these polymers. An understanding of the relationship between the chemical structure of the repeating unit of the polymer and its electrical properties would enable the electronic and mechanical properties of these materials to be tailored at the molecular level.

II. SYNTHESIS AND ANALYSIS OF COMPOSITE

The composite of the various combination of ratios the Pyrrole to FeCl₃ with the Fe₃O₄ nano particles are prepared. The composite thus prepared in the form of powder. The powder of composite is prepared through Chemical polymerization Method. The sample preparation starts from distillation of Pyrrole. With various combination of Pyrrole to FeCl₃ are stirred for overnight. Also the Fe₃O₄ nano particles are added as the blending purpose. The solution is then processed in centrifugal machine and filtered to separate the composite particles from the solution. The powder is dried in the furnace at warm temperature. Also the composite is exposed to room temperature for sufficient time. The dried powder sample are analyzed through XRD. FTIR of the same samples are done. The pellet are prepared from the powder using KBr press for further experimental process.

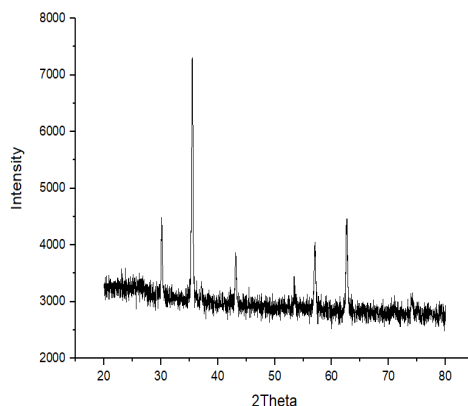


Figure 1: XRD for sample : Pyrrole (1ml) + FeCl₃ (1gm)+Fe₃O₄ (0.3gm)

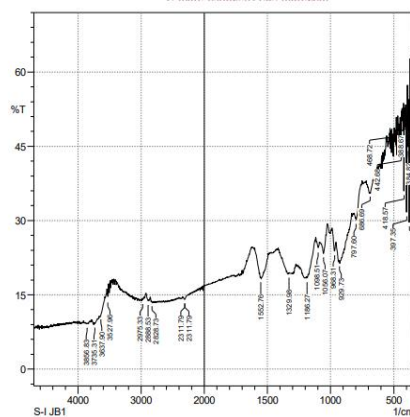


Figure 2: FTIR for Sample : Pyrrole (1ml) + FeCl₃ (1gm)+Fe₃O₄ (0.3gm)

III. CONCLUSION

The XRD pattern shown in Figure 1 gives information that the Peak are obtained at the 30.034°, 35.410°, 43.038°, 53.379° and 56.911° are similar that with obtained for pure Fe₃O₄ which confirm the presence of Fe₃O₄. The FTIR spectra of PPy and the Fe₃O₄–PPy composite is shown in Figure 2. PPy–Fe₃O₄ composites were prepared with the chemical polymerization method. The properties of the PPy–Fe₃O₄ particles were influenced by the concentration of FeCl₃ solution, which was used to disperse Fe₃O₄ particles.

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