

Study and Appropriate Analysis of Waste Tyre Carbon Black

**Sumit Ganguly, Divyanshu Kumar, Dinesh Mandal, Md. Naushad Ansari,
Sonu Kumar, Ajay Kr Singh and Santanu Pal**

Department of Mechanical Engineering
K. K. Polytechnic, Dhanbad, India

Abstract: *The amount of trash tires being disposed of is growing exponentially every day. In addition to taking up a significant quantity of precious ground, this waste tire poses a risk of a fire. An estimated 1.5 billion trash tires are produced annually worldwide, with China and India accounting for 41% of this total. Thermally reprocessing the tires into useful goods like activated carbon, various solid carbon forms (carbon black, graphite, and carbon fibers), and liquid fuels is a better way to use them more efficiently from an economic and environmental perspective. Pyrolysis is one of the heat processes that can turn used tires into solid, gas, or oil. With some quality enhancements, such as the removal of ash and sulfur, a solid product, for example, can be used as carbon black. Its use in a variety of goods is restricted by its high ash concentration. Numerous additives from the original tire contaminated the pyrolytic carbon black (CB) that was produced. Chemical reactions that took place in the pyrolysis reactor also produced contaminants. In order to maximize the benefits of activated carbon, we must separate the metal content from the carbon black that is produced by pyrolyzing waste tires*

Keywords: pyrolytic carbon black

I. INTRODUCTION

The amount of trash tires being disposed of is growing exponentially every day. In addition to taking up a significant quantity of precious ground, this waste tire poses a risk of a fire. An estimated 1.5 billion trash tires are produced annually worldwide, with China and India accounting for 41% of this total. These used tires are not adequately assessed. Only the United States, Japan, and the European Union manufacture roughly 6 million tons of discarded tires annually. As the related auto sectors expand, the enormous amount of discarded tires produced worldwide will rise in the future. The removal of used tires raises environmental concerns. Pollution of the environment results from the deposition of used tires. A significant portion of the used tires are merely disposed of in locations that pose risks for illnesses and unintentional fires. Rubber causes issues with disposal because it is not organically degradable. Recycling tires with material or energy recovery can reduce the amount of garbage they produce. Rubber materials cannot be easily softened and remolded by heating because they are cross-linked during manufacturing and molding. Landfills were the primary, workable solution to the waste tire issue for a many of years. Tire waste dumped in the open might lead to excessive pollution emissions and unintentional fires. In 1990, 62% of all tires produced in EU countries ended up in landfills; by 2000, that number had dropped to roughly 35%; and by 2006, as a result of EU law, the percentage would have dropped to 0%. The Slovak Republic is among the nations that have already outlawed the disposal of used tires in landfills [1]. Only a small portion of used tires are recycled and utilized again to make rubber goods. Waste tires are utilized as fuel in rotary cement kilns due to their high calorific value. Because of the harmful pollutants generated during tire burning, this procedure can only be justified from an environmental perspective in the situation of regulated combustion. We must recycle and repurpose these used tires in order to solve such issues. The pyrolysis process can be used to transform these waste tires into valuable items. The primary byproduct of pyrolyzing used tires is carbon black. The global availability of carbon black is depicted in Figure 1.

Waste Tyres All Over World

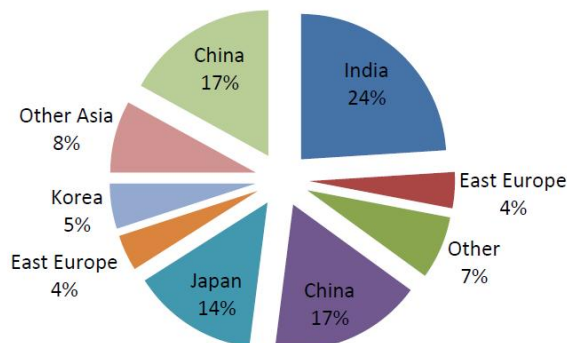


Fig.1. Waste Tyre [2]

Pyrolysis of scrap tires also yields other byproducts, such as steel wire, pyro oil, and pyro gas. Pyro oil and pyro gas can be utilized as secondary fuels, while steel wires can be used in the steel industry. Because of its high calorific value, the carbon black solid powder result of scrap tires can be transformed into activated carbon, which has potential applications.

II. METHODOLOGY

There are some methodologies for Analysis are below

PYROLYSIS OF WASTE TYRE

A thermochemical process called pyrolysis breaks down organic substances at a greater temperature. Pyrolysis is made up of the words "pyro" (fire) and "lysis" (breakdown). In the absence of oxygen, pyrolysis primarily involves the breakdown of organic matter into its most basic components. However, a commercial oxygen-free atmosphere is not achievable. Hydro-pyrolysis is the term for pyrolysis that occurs in the presence of water. The chemical industry makes extensive use of this method. Coke is a byproduct of the pyrolysis of coal; similarly, the pyrolysis process can yield many other beneficial products, such as charcoal and activated carbon from wood. Tipping fees for disposing of waste tires and the selling price of pyrolysis products are the two primary elements that impact the pyrolysis plant's profitability. The cost of tipping fees for disposing of waste is rising daily these days, but the poor market and calorific value of pyrolysis products are causing their selling prices to fall. Therefore, we must reprocess the products, such carbon black, in order to turn a profit. High-grade carbon black, activated carbon, and valuable compounds with a high calorific value that can be used as an alternative fuel can all be produced from carbon black. Pyrolysis is generally done between 500 and 900 degrees Celsius. Carbon black and the mineral materials that was originally in the tire are found in the solid residue [4]. This solid char can be utilized as smokeless fuel, activated carbon, or reinforcement in the rubber sector. A complex blend of organic components makes up the liquid product. As a result, the extracted oils can be utilized directly as fuels, feedstock for petroleum refineries, or as a source of chemicals. Non-condensable organics such as H₂, H₂S, CO, CO₂, CH₄, C₂H₄, C₃H₆, and others make up the majority of the gaseous portion. The pyrolysis process can be powered by the gas fraction.



Fig. 2 Setup of pyrolysis plant [3]

1, 2 – Both furnaces with oil fuel burners at front 3, 4 – both heating boxes with every of double-screw reactor inside; 5 – heat utilizing steam boilers.

The market for pyrolysis products determines the viability of pyrolysis as a waste tire recycling technique. Characterizing the byproducts of pyrolysis and using them in other processes is crucial for this. Nowadays, the primary uses of solid char are as smokeless fuel, active carbon, and reinforcement in the rubber sector. The gas portion can be utilized as fuel in the pyrolysis process, and the liquid output can be used as fuel or as a source of chemicals.

Components	%age present
Pyro-oil	35-45
Pyro-gas	25-15
Carbon-black	35-40
Steelwire	<10

ATOMIC ABSORPTION SPECTROSCOPY

It is a spectro analytical technique that uses radiation absorption by free atoms in a gaseous state to determine different chemical constituents of a sample. The technique of using an analyte's electromagnetic or mass spectrum to determine its elemental composition is called atomic spectroscopy.

There are numerous analytical methods available, and the secret to getting precise, trustworthy, practical results is choosing the best one. Making the right decision necessitates having a fundamental understanding of each approach because each has unique advantages and disadvantages. It also entails a careful consideration of the analytical needs of the laboratory.

There are three broadly accepted analytical methods – atomic absorption, atomic emission and mass spectrometry – which will form the focus of our discussion, allowing us to go into greater depth on the most common methods in use today:

- Flame Atomic Absorption Spectroscopy
- Graphite Furnace Atomic Absorption Spectroscopy
- Inductively Coupled Plasma Optical Emission Spectroscopy (ICP-OES)
- Inductively Coupled Plasma Mass Spectrometry (ICP-MS)

PROXIMATE ANALYSIS OF CARBON BLACK

By isolating its many components, such as volatile, fixed, and inert components, proximate analysis has been utilized to assess the rank of carbon black. There is a need for effective procedures due to the vast range of value of carbon black products and their commercial value. We can learn about the composition of carbon black by proximate analysis.

It mostly includes the following:

- (i) Moisture content
- (ii) Volatile matter content
- (iii) Ash content
- (iv) Fixed carbon content

MOISTURE CONTENT

The weight loss that occurs when a sample is dried in an oven to a fixed weight is used to calculate the moisture content. After being dried and weighed, around 2g of a feed sample is placed onto a silica dish. After being dried for 36 hours at 6500C in an oven, the sample is cooled in a desiccator and weighed. Until a consistent weight is reached, the drying and weighing process continues.

%Moisture = (weight of sample plus dish prior to drying)-(weight of sample plus dish following drying) × 100
weight of sample collected

Ingredients and feed are usually compared for their nutrient content on a moisture-free or dry matter (DM) basis because the water content of feed varies greatly. $100 - \% \text{Moisture} = \% \text{DM}$.

ASH CONTENT

Ash is the inorganic waste that is produced when feed material's organic matter is burned for four hours at 400–6000 degrees Celsius in a muffle furnace. A preheated crucible is filled with 2g of the sample. For four hours, or until white-gray ash is obtained, the crucible is placed in a muffle furnace set between 400 and 6000 degrees Celsius. After that, the crucible is placed inside the desiccator and weighed.

%Ash = (crucible weight + ash - crucible weight) ÷ sample weight

III. RESULT & DISCUSSION

In the institute lab, carbon black is analyzed using atomic absorption spectroscopy. Following the study, I discovered some metal in percentage form from the pyrolysis CB, which is shown in Table 2 below and in Table 3 for the proximal analysis.

Table 2

Name of Metal	Quantity Present(in mg/lit)
Nickel	0.062
Lead	0.088
Zinc	0.009

Table 3

Components	%
Moisture	1.66
Volatile matter	3.14
Ash	6.65
Fixed carbon	86.55

IV. CONCLUSION

It is abundantly evident from the aforementioned studies that carbon black has a very broad variety of applications. Various products derived from pyrolysis of waste tyre such as carbon black, pyro oil, pyro gas, steel wire etc. can be very beneficial in future. Secondary fuel sources include pyro oil and pyro gas. The steel industry can repurpose the obtained steel wire. Because of its high calorific value, carbon black is receiving greater attention these days. The

proximate analysis of carbon black revealed a high amount of carbon (86.55). As a result, it can be utilized as an alternate fuel and have a higher calorific value.

Traces of metals like lead, zinc, and nickel were discovered. Carbon black can be transformed into activated carbon once these metals have been removed. The market value of activated carbon is higher. The quantity of contaminants is decreased by this carbon black. Waste tires are discarded almost everywhere in the globe; the tyre pyrolysis process can be useful.

REFERENCES

- [1] Juma M, Koreňová Z, Markoš J, Jelemenský L', Annus J- Pyrolysis And Combustion Of Scrap Tire.
- [2] Ware Pundlik, Shukla Vikaskumar, Kushvah Avadhesh, Desai, Demineralization and Characterization of Carbon Black Obtained From Pyrol Tyre Using Thermal Shock Process International Journal of Research in Chemistry Tyre Using Thermal Shock Process International Journal of Research in Chemistry.
- [3] Uladzimir Kalitko -Waste Tire Pyrolysis Recycling with Steaming: Heat-Mass Balances & Engineering Solutions for By-Products Quality Heat-Mass Transfer Institute, HMTI, Belarus National Academy of Science, Minsk, Belarus
- [4] Hsisheng Teng, Michael A. Serio, Rosemary Bassilakis, Philip W. Morrison, Jr. and Peter R. Solomon- Reprocessing of used tires into activated carbon and other products- Advanced Fuel Research, Inc., 87 Church Street East Harford, CT 06108
- [5] Chaala A, Darmstadt H, Roy C- Acid-base method for the demineralization of pyrolytic carbon black- Fuel Processing Technology 46 (1996) 15
- [6] Jianfeng Huang, Fei Shen, Xianhui Li, Xuanquan Zhou, Binyao Lia, Renliang Xua, Chifei Wu- Chemical modification of carbon black by a simple non-liquid-phase approach
- [7] Shah Jasmin, Jan M. Rasul, Mabood Fazal and. Shahid M- Conversion of Waste Tyres into Carbon Black and their Utilization as Adsorbent- Journal of the Chinese Chemical Society, 2006, 53, 1085-1089