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Synthesis, Structural and Spectroscopic Characterization of a Ni(II) Dimeric Complex with a Tridentate NNO donor Schiff Base Ligand: Coexistence of Square Planar and Octahedral Geometries

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Abstract: A new Ni(II) complex, $[Ni_2L_2(OH)(OH_2)]ClO_4 \cdot CH_2Cl_2$ (1), was synthesized using a tridentate N,N,O-donor Schiff base ligand, HL, derived from the condensation of salicylaldehyde and N,N-dimethylethane-1,2-diamine. The complex was characterized by X-ray structural analysis and spectroscopic studies. The crystal structure reveals a Ni(II) dimer in which the two Ni atoms adopt distinct coordination environments: one in a square planar geometry and the other in an octahedral geometry. These Ni atoms are bridged by a phenoxo oxygen atom from the ligand and a hydroxo group.

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