

# Synthesis, Structural and Spectroscopic Characterization of a Ni(II) Dimeric Complex with a Tridentate NNO donor Schiff Base Ligand: Coexistence of Square Planar and Octahedral Geometries

Dr. Subrata Naiya<sup>1</sup> and Dr. Saptarshi Biswas<sup>2</sup>

Department of Chemistry, Sushil Kar College, Champahati, Baruipur, West Bengal, India<sup>1</sup>

Department of Chemistry, Katwa College, Katwa, Purba Bardhaman, West Bengal, India<sup>2</sup>

**Abstract:** A new Ni(II) complex,  $[\text{Ni}_2\text{L}_2(\text{OH})(\text{OH}_2)]\text{ClO}_4 \cdot \text{CH}_2\text{Cl}_2$  (**1**), was synthesized using a tridentate N,N,O-donor Schiff base ligand, HL, derived from the condensation of salicylaldehyde and N,N-dimethylethane-1,2-diamine. The complex was characterized by X-ray structural analysis and spectroscopic studies. The crystal structure reveals a Ni(II) dimer in which the two Ni atoms adopt distinct coordination environments: one in a square planar geometry and the other in an octahedral geometry. These Ni atoms are bridged by a phenoxo oxygen atom from the ligand and a hydroxo group.

**Keywords:** Tridentate Schiff base, Nickel(II), Crystal structure, Dinuclear complex