

International Journal of Advanced Research in Science, Communication and Technology (IJARSCT)

Volume 2, Issue 3, March 2022

## Synthesis, Characterization and Application of AlTiZrO<sub>4</sub> Nanomaterial

M. V. Patil<sup>1</sup>, S. R. Bamane<sup>2</sup>, S. M. Khetre<sup>3</sup> Department of Chemistry, Miraj Mahavidyalaya, Miraj, India<sup>1</sup> Sushila Shankarrao Gadave Mahavidyalaya, Khandala, Satara, India<sup>2</sup> Department of Chemistry, Dahiwadi College, Dahiwadi, Satara, India<sup>3</sup>

Correspondence Author: pramanikpatil@rediffmail.com<sup>1</sup>

Abstract: Oval shaped AlTiZrO<sub>4</sub> nanoparticles were synthesized by wet chemical co precipitation and muffle ignition method. The oval shapes of nanomaterial were confirmed using SEM imaging and spinal packing in crystals were determined on the basis of XRD spectrum. The surface functionalities over nanomaterial was confirmed using FTIR spectrum elucidating hydroxyl and oxide groups over surface for future water wet ability. Furthermore the porous nature and electronic states in nanomaterial were elaborated on the basis of UV-Vis. And PL spectral transitions along with matching SEM and XRD data. The very high porosity of this ceramic nanomaterial was confirmed by BET measurements and future water remediation applications were demonstrated using antimicrobial testing on Proteus Vulgaris and membrane water purification activity. Overall this novel ceramic porous nano material has proved probable application in water purification membranes.

Keywords: Oval ceramic, Nano material, Highly Porous, Water remediation, Absorbance

## REFERENCES

- [1]. Basheer A.A., Journal of Molecular Liquids, Nethelands, (2018), V.216, pp.583-593.
- [2]. Burggraaf, A. J., Keizer, K., Synthesis of Inorganic Membrane 35; (2009), 38-38.
- [3]. Buxton, G. V.; Greenstock, C. L.; Helman, W. P.; Ross, A. B. J. Phys. Chem. Ref. Data (1988), 17, 513.
- [4]. Connes, J. and Connes, P., Near Infrared Planetary spectra by Fourier Spectroscopy. I. Instruments and results.(1966) 56(7): 896-910.
- [5]. Egerton, R. F., Physical Principles of Election Microscopy : An Introduction to TEM, SEM and AEM. Springer, (2005) 202.
- [6]. Giamello, E.; Calosso, L.; Fubini, B.; Geobaldo, F. J. Phys. Chem. (1993), 97, 5735.
- [7]. Ghormely, J,A.; Stewart, A.C. J.Am. Chem. Soc.(1956), 78,2934.
- [8]. Haines, R. I.; McCracken, D. R.; Rasewych, J. B. In *Water Chemistry of Nuclear Reactor Systems 5*; British Nuclear Energy Society: London, (1989),p 309.
- [9]. Khali A., Gondal M.A., Dastageer M.A., Catalysis Communication, (2009), 11(3), pp.214-219.
- [10]. Li, W., Zhou, L.; Xing, W., Xu, N. Desalin. Water Treat. (2010), 18, 239-244.
- [11]. Marcells A., Omole, Issac K., Owino, Sadik, (2009), pp.233-247.
- [12]. Misra, P., Dubinski, M. eds., Ultraviolet Spectroscopy and UV Lasers. New York : Marcel Dekker (2002) ISBN 978-0-8267-0668-5.
- [13]. Savage, N., Diallo M., Nanomaterials and Water Purification: Opportunities and Challenges. Journal of Nanoparticle Research,(2005) 7, pp.331-342.
- [14]. Prabhu, V., Patwardhan, A. & Patwardhan, A.W., CIJCT). (2016) Vol.24, PP 367-373.
- [15]. Reimer, L., Scanning Electron Microscopy: Physics of Image formation and Microanalysis. (Springer), (1998), 527.P.
- [16]. The Infracord double-beam spectrophotometer. Clinical Science. (1957) 16 (2).
- [17]. Qu.X, Alvarez P.J.J., Li,Q., Water research, England, (2013), v.47, n.12, pp.3931-3946.
- [18]. Zhi-ChuanW., Yong Z., Zhangb L., Fongb H., Applied Surface Science, (2010),257(3),pp.1092-1097. Copyright to IJARSCT DOI: 10.48175/IJARSCT-3061

www.ijarsct.co.in

## IJARSCT



International Journal of Advanced Research in Science, Communication and Technology (IJARSCT)

Volume 3, Issue 3, March 2022