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Synthesis of Graphene Oxide Via Microwave Assisted Eco-Friendly Method Versus Modified Hummer Method

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Abstract: Using eco-friendly chemicals, a unique method for synthesizing graphene oxide (GO) by microwave radiation along with organic solvents is described. Microwave radiation was utilized to cause the graphite worm to expand significantly in all directions.SEM images were used to assess the morphology and structure of GO nanomaterials. X-ray diffraction, UV-Vis, and Fourier-transform infrared spectroscopies were used to identify the oxygenated groups on the GO surface. When compared to the conventional Hummer's technique, the new microwave-assisted approach significantly reduces reaction time and is much more eco-sustainable. Aside from being environmentally sustainable, the suggested approach has the appealing feature of producing a large number of graphene nano-sheets quickly. This article compares efficient and environmentally microwave-assisted and conventional modified hummer methods for graphene oxide preparation..

Keywords: Eco-friendly, Eco-sustainable, Microwave radiation, Nanomaterials, conventional.

REFERENCES

- Geim, A. K., &Novoselov, K. S., The rise of graphene. In Nanoscience and technology: a collection of reviews from nature journals, 11-19, 2010.
- [2]. Park, S., &Ruoff, R. S., Chemical methods for the production of graphenes. Nature nanotechnology, 4(4), 217-224, 2009.
- [3]. Morozov, S. V., Novoselov, K. S., Schedin, F., Jiang, D., Firsov, A. A., &Geim, A. K., Two-dimensional electron and hole gases at the surface of graphite. Physical Review B, 72(20), 201401, 2005.
- [4]. Bonaccorso, F., Sun, Z., Hasan, T., & Ferrari, A. C., Graphene photonics and optoelectronics. Nature photonics, 4(9), 611-622, 2010.
- [5]. Leary, R., & Westwood, A., Carbonaceous nanomaterials for the enhancement of TiO2 photocatalysis. Carbon, 49(3), 741-772, 2011.
- [6]. Chen, D., Feng, H., & Li, J., Graphene oxide: preparation, functionalization, and electrochemical applications. Chemical reviews, 112(11), 6027-6053, 2012.
- [7]. Jiang, F., Yao, Z., Yue, R., Du, Y., Xu, J., Yang, P., & Wang, C., Electrochemical fabrication of long-term stable Pt-loaded PEDOT/graphene composites for ethanol electrooxidation. International Journal of Hydrogen Energy, 37(19), 14085-14093, 2012.
- [8]. Yang, W., Ratinac, K. R., Ringer, S. P., Thordarson, P., Gooding, J. J., &Braet, F., Carbon nanomaterials in biosensors: should you use nanotubes or graphene?. AngewandteChemie International Edition, 49(12), 2114-2138, 2010.
- [9]. Lu, L., Zhang, O., Xu, J., Wen, Y., Duan, X., Yu, H., ... &Nie, T., A facile one-step redox route for the synthesis of graphene/poly (3, 4-ethylenedioxythiophene) nanocomposite and their applications in biosensing. Sensors and Actuators B: Chemical, 181, 567-574, 2013.

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- [10]. Gao, Y. S., Xu, J. K., Lu, L. M., Wu, L. P., Zhang, K. X., Nie, T., ... & Wu, Y., Overoxidizedpolypyrrole/graphenenanocomposite with good electrochemical performance as novel electrode material for the detection of adenine and guanine. Biosensors and Bioelectronics, 62, 261-267, 2014.
- [11]. Zhang, P., Zhang, X., Zhang, S., Lu, X., Li, Q., Su, Z., & Wei, G., One-pot green synthesis, characterizations, and biosensor application of self-assembled reduced graphene oxide–gold nanoparticle hybrid membranes. Journal of Materials Chemistry B, 1(47), 6525-6531, 2013.
- [12]. Wang, X., Huang, S., Zhu, L., Tian, X., Li, S., & Tang, H., Correlation between the adsorption ability and reduction degree of graphene oxide and tuning of adsorption of phenolic compounds. Carbon, 69, 101-112, 2014.
- [13]. Brodie, B. C., XIII. On the atomic weight of graphite. Philosophical transactions of the Royal Society of London, (149), 249-259, 1859.
- [14]. Staudenmaier, L., Verfahrenzurdarstellung der graphitsäure. Berichte der deutschenchemischen Gesell schaft, 31(2), 1481-1487, 1898.
- [15]. Hummers Jr, W. S., &Offeman, R. E., Preparation of graphitic oxide. Journal of the american chemical society, 80(6), 1339-1339, 1958.
- [16]. Park, S., &Ruoff, R. S., Chemical methods for the production of graphenes. Nature nanotechnology, 4(4), 217-224, 2009.
- [17]. Park, S., &Ruoff, R. S., Chemical methods for the production of graphenes. Nature nanotechnology, 4(4), 217-224, 2009.
- [18]. Vadukumpully, S., Paul, J., & Valiyaveettil, S., Cationic surfactant mediated exfoliation of graphite into graphene flakes. Carbon, 47(14), 3288-3294, 2009.
- [19]. Zhamu, A., & Jang, B. Z., U.S. Patent No. 8,753,539. Washington, DC: U.S. Patent and Trademark Office, 2014.
- [20]. Viana, M. M., Lima, M. C., Forsythe, J. C., Gangoli, V. S., Cho, M., Cheng, Y., ... & Caliman, V., Facile graphene oxide preparation by microwave-assisted acid method. Journal of the Brazilian Chemical Society, 26, 978-984, 2015.
- [21]. Li, D., Müller, M. B., Gilje, S., Kaner, R. B., & Wallace, G. G., Processable aqueous dispersions of graphene nanosheets. Nature nanotechnology, 3(2), 101-105, 2008.
- [22]. Ban, F. Y., Majid, S. R., Huang, N. M., & Lim, H. N., Graphene oxide and its electrochemical performance. Int. J. Electrochem. Sci, 7(5), 4345-4351, 2012.